

CHEM1047 – Week 3 Lecture 2 – Partial derivatives

- Section 9.3 of Steiner, "The Chemistry Maths Book", 2nd edition.
- Section 4.5 of Cockett and Doggett, "Maths for Chemistry", Volume 1.

1. Partial derivatives

A *partial derivative* of a function is a derivative with respect to just one of its arguments. The same differentiation rules apply, all other arguments should be treated as constants. Examples:

$$\frac{\partial}{\partial y}(xyz) = xz \qquad \frac{\partial}{\partial x}\left(\frac{1-x}{\sin y}\right) = \frac{-1}{\sin y}$$

Higher derivatives, such as $\partial^2 f / \partial x^2$, and mixed derivatives, such as $\partial^2 f / \partial x \partial y$, are obtained by sequential partial differentiation. For well-behaved functions, the value of mixed derivatives does not depend on the order of partial differentiation.

Partial derivatives have the same limit definition as the derivative of a univariate function:

$$\frac{\partial f(x, y, z, \dots)}{\partial x} = \lim_{\Delta x \rightarrow 0} \frac{f(x + \Delta x, y, z, \dots) - f(x, y, z, \dots)}{\Delta x} \quad (1)$$

The arguments are often collected into a vector for convenience, *e.g.*:

$$f(x, y, z) = f(\vec{r}), \qquad \vec{r} = [x \quad y \quad z] \quad (2)$$

Example 1:

$$z = \sin(xy)$$

$$\frac{\partial z}{\partial x} = y \cos(xy) \qquad \frac{\partial z}{\partial y} = x \cos(xy) \qquad \frac{\partial^2 z}{\partial x^2} = -y^2 \sin(xy)$$

$$\frac{\partial^2 z}{\partial x \partial y} = \frac{\partial}{\partial y}\left(\frac{\partial z}{\partial x}\right) = \frac{\partial}{\partial y}[y \cos(xy)] = \cos(xy) - xy \sin(xy)$$

Example 2:

$$z = x^2 + y^2$$

$$\frac{\partial z}{\partial x} = 2x \qquad \frac{\partial z}{\partial y} = 2y \qquad \frac{\partial^2 z}{\partial x^2} = 2 \qquad \frac{\partial^2 z}{\partial x \partial y} = 0$$

2. Chain rules for multivariate functions

In situations when a function depends on multiple arguments that are themselves functions of other arguments, the chain rule for differentiation must be extended appropriately. Several situations exist:

1. A function of one variable that depends on multiple other variables:

$$\frac{\partial}{\partial x} f[g(x, y, z, \dots)] \quad (3)$$

We can use the definition of the derivative twice to write out Δf when the argument x is incremented by Δx , and then take the limit:

$$\Delta f = \frac{df}{dg} \Delta g + O[\Delta g^2] = \frac{df}{dg} \left(\frac{\partial g}{\partial x} \Delta x + O[\Delta x^2] \right) + O[\Delta g^2] \quad (4)$$

$$\frac{\partial f}{\partial x} = \lim_{\Delta x \rightarrow 0} \frac{\Delta f}{\Delta x} = \frac{df}{dg} \frac{\partial g}{\partial x}$$

We therefore arrive at the following chain rule:

$$\frac{\partial}{\partial x} f[g(x, y, z, \dots)] = \frac{df}{dg} \frac{\partial g}{\partial x} \quad (5)$$

2. A function of multiple variables that each depend on the same variable:

$$\frac{d}{dt} f[x(t), y(t), \dots] \quad (6)$$

This is a full derivative because the function ultimately only depends on one argument. Using the definition of a partial derivative followed by the definition of univariate derivative, we get:

$$\begin{aligned} \Delta f &= \frac{\partial f}{\partial x} \Delta x + \frac{\partial f}{\partial y} \Delta y + \dots + O[\Delta x^2] + O[\Delta y^2] + \dots = \\ &= \frac{\partial f}{\partial x} \frac{dx}{dt} \Delta t + \frac{\partial f}{\partial y} \frac{dy}{dt} \Delta t + \dots + O[\Delta x^2] + O[\Delta y^2] \end{aligned} \quad (7)$$

After using the limit definition of the derivative with respect to t , we can conclude that:

$$\frac{d}{dt} f[x(t), y(t), \dots] = \frac{\partial f}{\partial x} \frac{dx}{dt} + \frac{\partial f}{\partial y} \frac{dy}{dt} + \dots \quad (8)$$

3. A function of multiple variables that each depend on multiple other variables. This is a combination of the two situations above and the corresponding rule may be derived in a similar fashion:

$$\frac{\partial}{\partial \alpha} f[x(\alpha, \beta, \dots), y(\alpha, \beta, \dots), \dots] = \frac{\partial f}{\partial x} \frac{\partial x}{\partial \alpha} + \frac{\partial f}{\partial y} \frac{\partial y}{\partial \alpha} + \dots \quad (9)$$

Example 3:

$$z = \sin(x), \quad x = u^2 v^2$$

$$\frac{\partial z}{\partial u} = \frac{dz}{dx} \frac{\partial x}{\partial u} = [\cos(x)] [2uv^2] = 2uv^2 \cos(u^2 v^2)$$

Example 4:

$$z = x^2 y - y^2, \quad x = t^2, \quad y = 2t$$

$$\frac{dz}{dt} = \frac{\partial z}{\partial x} \frac{dx}{dt} + \frac{\partial z}{\partial y} \frac{dy}{dt} = [2xy] [2t] + [x^2 - 2y] [2] = \dots = 10t^4 - 8t$$

Example 5:

$$z = e^{x^2 y}, \quad x(u, v) = \sqrt{uv}, \quad y(u, v) = 1/v$$

$$\frac{\partial z}{\partial u} = \frac{\partial z}{\partial x} \frac{\partial x}{\partial u} + \frac{\partial z}{\partial y} \frac{\partial y}{\partial u} = [2xye^{x^2 y}] \left[\frac{\sqrt{v}}{2\sqrt{u}} \right] + [x^2 e^{x^2 y}] [0] = \dots = e^u$$

3. Gradient vector

An important theorem in physics states that the force acting on a particle freely moving in some potential is equal to the partial derivatives of that potential:

$$E = U(x, y, z) \quad \Rightarrow \quad \vec{F} = \left[\frac{\partial U}{\partial x} \quad \frac{\partial U}{\partial y} \quad \frac{\partial U}{\partial z} \right] \quad (10)$$

Because force is a vector, it is often convenient to store all partial derivatives of a function in a vector as shown in Equation (10). Such a vector is called the *gradient* of the function:

$$\vec{\nabla}U = \left[\frac{\partial U}{\partial x} \quad \frac{\partial U}{\partial y} \quad \frac{\partial U}{\partial z} \quad \dots \right] \quad (11)$$

Gradient indicates the direction of steepest growth of the function and has many uses in physical sciences – for example, one of the best methods for decreasing the value of the function (called *gradient descent*) is to move in the direction opposite to its gradient.

4. Hessian matrix

Consider a weight of mass m attached to a spring of stiffness k and equilibrium length x_0 . From elementary mechanics we know that the total energy is the sum of kinetic and potential energies:

$$E = \frac{mv^2}{2} + \frac{k(x-x_0)^2}{2} \quad (12)$$

For a molecule with multiple atoms connected by multiple bonds, the expression is similar:

$$E = \frac{1}{2} \sum_n m_n |\vec{v}_n|^2 + \frac{1}{2} \sum_{n>m} k_{nm} (|\vec{r}_{nm}| - |\vec{r}_{nm}^{\text{eq}}|)^2 \quad (13)$$

where m_n is the mass of n -th atom, \vec{v}_n is its velocity vector, \vec{r}_{nm} is the distance vector between atom n and atom m , and the “eq” superscript refers to the equilibrium value. It is easy to see that the second derivatives of the energy with respect to the inter-atomic distances are the force constants:

$$\frac{\partial^2 E}{\partial |\vec{r}_{nm}|^2} = k_{nm} \quad (14)$$

This is very useful in practice because it allows vibrational frequencies to be calculated from derivatives of the total energy. The matrix of second derivatives of a function, for example:

$$\mathbf{H}[f(x, y)] = \begin{pmatrix} \frac{\partial^2 f}{\partial x^2} & \frac{\partial^2 f}{\partial x \partial y} \\ \frac{\partial^2 f}{\partial y \partial x} & \frac{\partial^2 f}{\partial y^2} \end{pmatrix} \quad (15)$$

is called the *Hessian matrix*. It contains information about the local curvature of the function.