Chromophore Structure in the Photocycle of the Cyanobacterial Phytochrome Cph1

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ABSTRACT The chromophore conformations of the red and far red light induced product states “Pfr” and “Pr” of the N-terminal photoreceptor domain Cph1-NS15 from Synechocystis 6803 have been investigated by NMR spectroscopy, using specific 13C isotope substitutions in the chromophore. 13C-NMR spectroscopy in the Pfr and Pr states indicated reversible chemical shift differences predominantly of the C4 carbon in ring A of the phycocyanobilin chromophore, in contrast to differences of C15 and C5, which were much less pronounced. Ab initio calculations of the isotropic shielding and optical transition energies identify a region for C4-C5-C6-N2 dihedral angle changes where deshielding of C4 is correlated with red-shifted absorption. These could occur during thermal reactions on microsecond and millisecond timescales after excitation of Pr which are associated with red-shifted absorption. A reaction pathway involving a hula-twist at C6 could satisfy the observed NMR and visible absorption changes. Alternatively, C15-ZE photoisomerization, although expected to lead to a small change of the chemical shift of C15, in addition to changes of the C4-C5-C6-N2 dihedral angle could be consistent with visible absorption changes and the chemical shift difference at C4. NMR spectroscopy of a 13C-labeled chromopeptide provided indication for broadening due to conformational exchange reactions in the intact photoreceptor domain, which is more pronounced for the C- and D-rings of the chromophore. This broadening was also evident in the F2 hydrogen dimension from heteronuclear 1H-13C HSQC spectroscopy, which did not detect resonances for the 13C4-H, 13C10-H, and 13C16-H hydrogen atoms whereas strong signals were detected for the 13C-labeled chromopeptide. The most pronounced 13C-chemical shift difference between chromopeptide and intact receptor domain was that of the 13C4-resonance, which could be consistent with an increased conformational energy of the C4-C5-C6-N2 dihedral angle in the intact protein in the Pr state. Nuclear Overhauser effect spectroscopy experiments of the 13C-labeled chromopeptide, where chromophore-protein interactions are expected to be reduced, were consistent with a ZZZssa conformation, which has also been found for the biliverdin chromophore in the x-ray structure of a fragment of Deinococcus radiodurans bacteriophytochrome in the Pr form.

INTRODUCTION

Phytochromes are red and far red light receptors in plants and cyanobacteria that have various physiological roles (2,3). The fundamental spectroscopic changes, which are associated with receptor activation, are similar in most kinds of phytochromes. A red light (~650 nm) absorbing state, called “Pr”, is transformed with relatively low quantum yield (10%) into a far red-absorbing “Pfr” form, which can be retransformed with similar quantum yield using far red light (~710 nm) (4–8). An exception is the biliverdin-containing bacteriophytochrome photoactive yellow protein-phytochrome related from Rhodospirillum centenum, which has strongly overlapping Pr and Pfr absorption spectra with maxima at 702 nm but with a lower extinction for the Pfr state (9). The Pr states of most phytochromes and bacteriophytochromes (Bphs) are thermally the most stable forms, as has also been found for the cyanobacteriophytochrome Cph1 from Synechocystis 6803 (10). This observation has been cited in relation to the expected ZZZ (C5=C10=C15) conformation of all three bridging carbon atoms of the linear tetrapyrrole chromophores of phytochromes (11). In particular, free tetrapyrrole compounds such as phyco-cyanobolin are known to adopt helical ZZZ conformations in solution (12–16). However, the biliverdin-containing bacteriophytochrome AtBphP2 from Agrobacterium tumefaciens thermally relaxes to a Pfr-like ground state in the dark (17), as do also other bacteriophytochromes (18,19), indicating that the lowest energy conformation available to the free tetrapyrrole chromophores does not necessarily dictate the thermally most stable conformation when bound to phytochrome light receptors.

The phototransformation from the Pr state to the Pfr state has been proposed to involve a Z→E isomerization at the C15=C16 bond between the C- and D-rings of the linear tetrapyrrole chromophore (20–22). Time-resolved and low temperature trapping experiments are consistent with an initial photoisomerization reaction of both Pr and Pfr, followed by a number of slow thermal reactions. Cph1 shows optical and kinetic properties which are representative for many phytochromes, which include a slightly red-shifted lumir-R photoprotein of Pr formed 100 ps after excitation (8). Five subsequent kinetic components are observable on slower timescales (τ1−τ5: 5 and 300 μs and 3, 30, and 300 ms), which together are responsible for the red-shifted absorption of the Pfr product state (7). Similarly, low temperature trapping of the initial photoprotein lumir-R of Pr below 210 K produced less red-shifted absorption compared to the Pfr state that is produced at high temperature (23). Transient and steady-state protonation studies showed that the chromophore is fully

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protonated in both Pr and Pfr states (7). Therefore, the thermal transformations producing red-shifted products which occur on microsecond and millisecond timescales are likely to result from chromophore configurational changes, additionally considering that many phytochromes show similar spectroscopic and kinetic properties despite having different amino acid sequences.

NMR experiments suggested that a chromopeptide prepared from oat phytochrome in the Pr form has the \( \text{C}_{15}-\text{E} \) configuration, whereas a peptide derived from the Pr form has a \( \text{C}_{15}-\text{Z} \) configuration (20). Resonance Raman spectroscopy has identified a strong peak of Pr at 820 cm\(^{-1} \) belonging to the \( \text{C}_{15}-\text{H} \) hydrogen out of plane mode, which was argued to be consistent with a nonplanar conformation of the \( \text{C-} \) and \( \text{D-} \) rings in the Pfr state and supporting the \( \text{C}_{15}=\text{C}_{16} \text{ Z} \to \text{E} \) isomerization (21). A similar mode was identified in the spectra of Cph1 as well, suggesting the same reaction model (22). Calculations of Raman frequencies and intensities of molecular models of the phytochromobilin chromophore of oat phytochrome have refined this reaction model further and invoke an initial \( \text{ZZZ} \)asa (\( \text{C}_{3}-\text{Z},\text{C}_{10}-\text{Z},\text{C}_{15}-\text{C},\text{C}_{10}^\text{anti},\text{C}_{10}^\text{syn},\text{C}_{15}^\text{anti} \)) to \( \text{ZZE} \)asa photoisomerization of Pr transition to the lumi-R-photointermediate, followed by a partial thermal \( \text{ZZE} \)asa to \( \text{ZZE} \)asa \( \text{C}_{5}-\text{C}_{6} \) bond rotation producing the Pfr state (24–26). Recently the x-ray structure of a fragment of \( \text{Deinococcus radiodurans} \) bacteriophytochrome \( \text{DrBphP} \) was reported in the Pr state with the biliverdin chromophore modeled in the \( \text{ZZZ} \)asa conformation (1). Evidence for \( \text{C}_{15} \text{ Z-E} \) photoisomerization from this structure includes the proximity between Tyr-167 and the \( \text{D-} \) ring of the chromophore, which in the homologous cyanobacterial phytochrome Cph1 was shown to abolish Pr phototransformation and increase the fluorescence quantum yield when mutated to histidine (27). Here, we use \( ^{13}\text{C} \) direct detection NMR spectroscopy of cyanobacterial phytochrome Cph1 with \( ^{13}\text{C} \)-labeled phycocyanobilin chromophore to probe the structural changes associated with the Pr to Pfr transition and discuss reaction models that would be consistent with the nuclear magnetic shielding and transient and stable absorption changes.

**MATERIALS AND METHODS**

**Sample preparation and NMR spectroscopy**

A fragment containing the 515 N-terminal amino acid residues of Cph1 from \( \text{Synechocystis} \) 6803 (Cph1-N515), kindly provided by J. Clark Lagarias, was expressed together with heme oxygenase and bilin reductase, as previously described (28), following a similar procedure (29). A \( \text{hema} \) aminolevulinic acid auxotrophic BL21 (DE3) strain lacking the glutamyl-tRNA reductase gene (30) was used together with 0.5 mM 5-aminolevulinic-5-\( ^{13}\text{C} \) acid (Isotec, Miamisburg, OH) in the expression medium for the expression of \( ^{13}\text{C} \)-labeled phycocyanobilin chromophore to probe the structural changes associated with the Pr to Pfr transition and discuss reaction models that would be consistent with the nuclear magnetic shielding and transient and stable absorption changes.

A molecular model for the phycocyanobilin chromophore in the \( \text{ZZZ} \)asa geometry was taken from the 1.45-Å resolution x-ray structure of C-Phycocyanin from \( \text{Synechococcus elongates} \), PDB 1JBO (35) from the protein data bank. (36) A \( \text{ZZZ} \)asa phycocyanobilin model was based on the \( \text{ZZZ} \)asa biliverdin structure of the \( \text{D. radiodurans} \) bacteriophytochrome fragment, PDB 1ZTU (1). The sulfur linkage was replaced with a hydrogen atom, and all pyrrole nitrogen atoms were protonated. The propionate carboxyl groups were replaced with hydrogen atoms. All calculations were performed using Gaussian 03 (37). In vacuo density functional theory (DFT) (38,39) geometry optimization calculations, gauge including atomic orbital (GIAO) isotropic chemical shielding calculations (40–43), and time-dependent DFT (TD-DFT) excited state calculations (44,45) of the cation models were all performed at the DFT MP21PW91 6-31G(d,p) level (46). All isotropic shielding calculations are given relative to the values calculated for tetramethylsilane (TMS) calculated at the same level of theory. TD-DFT results given are the lowest lying transition energies with significant oscillator strengths, which in all cases provided the isolated HOMO-LUMO transition.

**RESULTS**

\( ^{13}\text{C} \)-NMR direct detection of labeled chromophore in intact Cph1-N515 and chromoprotein

Of the \( \text{C}_{4}, \text{C}_{5}, \text{C}_{6}, \text{C}_{10}, \text{C}_{11}, \text{C}_{15} \), and \( \text{C}_{19} \) carbon atoms replaced with \( ^{13}\text{C} \) isotopes, only the bridging \( \text{C}_{5}, \text{C}_{10} \), and \( \text{C}_{15} \) methine carbons have hydrogen atoms attached (Fig. 1). \( ^{13}\text{C} \)
direct detection of intact protein in the mixed Pr and Pfr state with labeled chromophore showed a series of broad peaks in addition to the broad envelope of superimposed signals at natural abundance of the unlabeled 58-kDa polypeptide. For the intact protein the peak widths were 30–40 Hz, typical of resonances of high molecular mass compounds, whereas chromopeptide peak widths were 15 Hz. From the expected dominant \(^{13}\)C–\(^1\)H dipolar interactions, contributions to the line widths could correspond to rotational correlation times in the order of 30 ns and 12 ns, respectively (47). Comparison of the \(^{13}\)C-spectra of unlabeled and labeled protein was required to unambiguously identify the peaks belonging to the phycocyanobilin chromophore (Fig. 1A). Assignment of the peaks was aided by isotropic chemical-shielding calculations (see below) and substantiated by the multiplicity and \(^1\)J\(^{13}\)C–\(^1\)H-coupling analysis, which was in agreement with local bond orders (Table 1) and previously published assignments of bilin compounds (14).

**The chromophore is in intermediate conformational exchange in both the Pr and Pfr forms in the intact Cph1-N515 sample**

\(^1\)H–\(^{13}\)C HSQC spectra of isotopically labeled Cph1-N515 failed to show crosspeaks for the \(^{13}\)C\(_5\)-H\(_1\), \(^{13}\)C\(_{10}\)-H\(_2\), and \(^{13}\)C\(_{15}\)-H-chromophore hydrogens, in agreement with a recent study (48). The possibility of paramagnetic contamination was excluded from electron paramagnetic resonance spectroscopy at cryogenic temperature and by proton-induced x-ray emission (MicroPIXE) measurements (not shown). Aggregation was similarly excluded from the line widths of \(^1\)H-NMR spectra. In the range between 100 \(\mu\)M and 1 mM, no significant changes in line widths of the \(^1\)H-NMR spectra of Chp1-N515 were observed, which could be characteristic of a monomeric, or of a rapidly exchanging

| \(^{13}\)C-NMR parameters for the intact protein and the chromopeptide |
|----------------------------------|---------------------|---------------------|
| **Intact** (mult; \(J_{CC}\)/ppm/Hz) | **Chromopeptide** (mult; \(J_{CC}\)/ppm/Hz) | **Chromopeptide Intact/ppm** |
| \(^{13}\)C\(_{19}\) | 175.3 (s) | 177 (s) | +1.5 |
| \(^{13}\)C\(_{4}\) (Pfr) | 151.9 (d; 65) | 152.9 (d; 81) | +4.0 |
| \(^{13}\)C\(_{4}\) (Pr) | 148.9 (d; 74) | 133.8 (d; 72) | -0.9 |
| \(^{13}\)C\(_9\) | 134.7 (d; 70) | 132.6 (d; 70) | -0.9 |
| \(^{13}\)C\(_{10}\) | 133.5 (d; 61) | 118.8 (t; 71) | +3.0 |
| \(^{13}\)C\(_{15}\) | 96.5 (s) | 98.5 (s) | +2.0 |
| \(^{13}\)C\(_{5}\) | 94 (d; 60) | 92.5 (d; 79) | -1.5 |

\(^{13}\)C-NMR chemical shifts are reported relative to TMS. Multiplicities and \(J_{CC}\) values are given in brackets.
dimeric, 58-kDa polypeptide. For both Pr and Pfr, concentrations used were in excess of homodimerization dissociation constants reported, (49) but line widths were less than expected for the rigid dimer. Therefore no light-induced changes of populations are expected, which is also corroborated by the similar $^{13}$C line widths in both Pr and Pfr states (Fig. 1). After trypsin digestion of the $^{13}$C-labeled receptor, the same sample at identical concentration showed strong doublet, triplet, and singlet peaks, for the $^{13}$C$_5$-H-, $^{13}$C$_{10}$-H-, and $^{13}$C$_{15}$-H-protons, with $^1$H chemical shifts of 5.55, 7.22, and 6.15 ppm, respectively, whereas no peaks were observed in unlabeled material at the same concentration (Fig. 2, B and D). The multiplicity and $^1$J$_{CC}$ couplings of the observed peaks matched those that were determined from the $^{13}$C-NMR experiments (Figs. 1 and 2; Table 1). The one-dimensional $^{13}$C-spectra showed that the resonances of, in particular, C$_{15}$ and C$_{10}$ increased multiplefold in intensity relative to the peptide peaks upon digestion with trypsin (Fig. 2, A and C). This was also observed for the C$_4$, C$_5$, C$_9$, C$_{11}$, and C$_{19}$ peaks (not shown) and was most pronounced for the C$_{15}$ and least pronounced for the C$_5$ peak (Fig. 2 A). $^1$H and $^1$H-$^{15}$N TROSY-HSQC spectra of intact $^{15}$N globally labeled Cph1-N515 at the same concentration were characteristic of a 58-kDa monomeric polypeptide, but the heteronuclear experiment showed considerable broadening of selected resonances (not shown). Together, the data are consistent with the presence of equilibrium conformational exchange reactions in the chromophore in the intact protein on the timescale of

**FIGURE 2** Conformational exchange reactions of the chromophore in the intact protein. (A) $^{13}$C-NMR spectra of the $^{13}$C$_5$ and $^{13}$C$_{15}$ carbons in Cph1-N515 (solid line) and chromopeptide (dashed line) under identical experimental conditions and concentration. Arrows indicate chromophore peaks. (B) $^1$H-$^{13}$C HSQC crospeaks for the $^{13}$C$_5$-H and $^{13}$C$_{15}$-H protons in the labeled chromopeptide. No peaks were observed in this region in unlabeled chromopeptide under identical conditions. (C) $^{13}$C-NMR spectra of the $^{13}$C$_{10}$ triplet in Cph1-N515 (solid line) and chromopeptide (dashed line), indicated with arrows. (D) $^1$H-$^{13}$C HSQC crospeaks of the $^{13}$C$_{10}$-H triplet in the labeled chromopeptide (light contours) shown together with the unlabeled chromopeptide, which shows contributions from superimposed resonances at natural abundance in this region (dark contours).
the carbon and proton or nitrogen resonance frequency changes accompanying the reaction. The exchange effect was notable in the $^{13}$C-spectra and therefore inferred in $^1$H and $^{15}$N experiments. This broadening of the chromophore resonances in the intact protein is less pronounced proximal to the covalent attachment site at C3', indicating that the change in frequency of these resonances is smaller, leading to faster exchange.

**The chromopeptide-bound phycocyanobilin chromophore is in the ZZZssa conformation**

To determine the molecular geometry of the phycocyanobilin chromophore in the chromopeptide, NOESY spectra were recorded on the $^{13}$C-labeled material. Crosspeaks were seen between the C$_5$-$^1$H-resonance, observed in these experiments at 5.68 ppm, and peaks at 2.02, 3.18, and 3.35 ppm, assigned to C$_7$-$^1$H, C$_3$-$^1$H, and C$_3'$-$^1$H, respectively (Fig. 3). The C$_{15}$-$^1$H-resonance at 6.29 ppm showed a strong crosspeak to a resonance at 2.14 ppm, assigned either to C$_{13}$-$^1$H or C$_{17}$-$^1$H (Fig. 3). The 2.14-ppm resonance consists of two closely spaced peaks, with a separation of 8 Hz (Fig. 3), very similar to the peak shapes of the three methyl resonances at 2.02, 2.08, and 2.14 ppm in purified phycocyanobilin in pyridine (not shown). The origin of this peak doubling is uncertain but could be due to gauche and anti-gauche conformations and is observed in purified phycocyanobilin as well as in the chromopeptide spectra. The chromopeptide spectra therefore did not indicate heterogeneity beyond that observed for the purified chromophore.

**Pr to Pfr phototransformation results in decreased shielding of C$_4$**

Illumination of Cph1-N515 with far red light produces the pure Pr state, which is the stable form in the dark. Subsequent illumination of NMR samples in thin capillaries with 640-nm light re-forms the Pfr state, which is metastable for several days under conditions used for $^{13}$C direct detection (see Materials and Methods; (10)). Repeated Pr$\rightarrow$Pfr and Pfr$\rightarrow$Pr phototransformations confirmed reversible changes in the frequency of the $^{13}$C$_4$-carbon resonance (Fig. 4). In the mixed Pr/Pfr state two doublets are visible for the $^{13}$C$_4$-carbon, at 151.9 and 148.9 ppm, respectively, whereas after illumination with far red light, only the doublet at 148.9 ppm is observed and increases in intensity (Fig. 4). Changes of the other peaks are much less pronounced. A possible reduction in intensity and perhaps change in frequency of the $^{13}$C$_5$-doublet at 94 ppm is observed, whereas no change is observed for the frequency of the $^{13}$C$_{15}$-resonance at 96 ppm. A small reduction in the intensities of peaks belonging to $^{13}$C$_9$, $^{13}$C$_{10}$, $^{13}$C$_{11}$, $^{13}$C$_{15}$, and possibly $^{13}$C$_{19}$ in the Pr state relative to the Pfr state (not shown) may indicate a change in conformational dynamics.

**DISCUSSION**

**Conformational exchange reactions and protein interaction of the chromophore**

The conformational exchange reactions of the intact photoreceptor domain hampered $^1$H-$^{13}$C HSQC spectroscopy, but resonances for the seven carbons labeled with $^{13}$C could be observed by $^{13}$C-NMR spectroscopy. Line broadening was observed beyond that expected from slow tumbling, manifested by a much reduced intensity relative to chromopeptide resonances at the same concentration. MicroPIXE (50) measurements and ESR spectroscopy at ambient and cryogenic temperatures confirmed that the broadening in the intact protein is not caused by paramagnetic contamination, and aggregation of the intact receptor was also excluded from $^1$H-NMR spectra. Additionally, the increased broad-
ening was removed by proteolytic digestion of the intact material. Hence, conformational exchange reactions are occurring that affect the line width mostly in the $^1$H- but also in the $^{13}$C-frequency domain.

Trypsin digestion removed conformational exchange broadening of the chromophore resonances and strong $^1$H-$^{13}$C HSQC crosspeaks were subsequently observed for the C_5-H, C_{10}-H, and C_{15}-H chromophore atoms (Fig. 2). The pronounced gain of intensity of, in particular, the $^{13}$C_{15}-carbon resonance after trypsin digestion suggests that the exchange broadening in the intact protein is greatest at the D-ring end of the molecule. This indicates that the conformational change results in larger changes in chemical shift at the D-ring (these resonances are thus in intermediate exchange) and smaller changes in chemical shift at the A-ring end of the molecule (thus in faster exchange). We note that the binding site proximal half, comprising rings A and B, with the exception of C_4 become deshielded upon interacting with the protein, whereas the distal end, comprising rings C and D, becomes more shielded (Table 1). The chromopeptide sample contains 10% DMSO-$_d^8$, which may contribute to some of the chemical shift changes, but the general trend is noteworthy.

We speculate that aromatic stacking on the distal end of the chromophore causes the shielding effect. The conformational exchange of the chromophore observed by NMR spectroscopy may be directly related to the temperature dependence of fluorescence of Cph1, which was interpreted to reflect conformational heterogeneity (51). Additionally, multiple decay phases of the picosecond absorption changes with excitation of the Pr state of Cph1 was interpreted in terms of heterogeneity, or substates, by fitting a distribution of rate constants to the data (8). Conformational exchange reactions of parts of the polypeptide was also observed by $^1$H-$^{15}$N TROSY-HSQC spectroscopy of uniformly $^{15}$N-labeled intact Cph1-N515, which showed broadening of a substantial portion of the amide resonances (not shown). The slow conformational exchange reactions which are occurring in the intact material, but not in the digested material, strongly affect the NMR spectroscopy observations and to some extent possibly also the optical properties. The occurrence of closely lying ground state conformations which are separated by thermal barriers result from chromophore-protein interactions which may also affect the phototransformation properties of the intact receptor.

**ZZZssa chromophore conformation in the chromopeptide**

The observed NOESY peaks from the C_5-H proton in the $^{13}$C-labeled chromopeptide dictate a C_4-Z, C_5-syn conformation, which fixes the relative positions of rings A and B. The observed single nuclear Overhauser effect (NOE) between C_{15}-H and either, but not both, the methyl protons on rings C or D dictates either a C_{14}-anti, C_{15}-Z or a C_{14}-syn, C_{15}-E conformation. Considering the possible C_{14}-syn, C_{15}-E structures, a ZZEsss conformation is not possible for sterical reasons, but a ZEEesas structure could be consistent with the NOE data. Considering possible C_{14}-anti, C_{15}-Z structures, the ZZZssa is most likely to be the lowest energy conformation. The ZZZssa and ZEEesas geometries were optimized using DFT and found to be, respectively, 17 and 67 kJ/mol higher in energy than the most stable ZZZssa conformation for the fully protonated state. Therefore, the ZZZssa conformation is most likely to exist in the chromopeptide, where stabilizing protein interactions are expected to be reduced. This conformation is tentatively supported by the ZZZssa biliverdin conformation, which was found in the x-ray structure of the homologous bacteriophytochrome fragment (1). The ZZZssa structure of the chromopeptide-bound phycocyanobilin at low pH deviates significantly from the helical ZZZsss conformation found for the purified chromophore (14,16). $^1$H, $^1$H-NOE enhancements were reported for C_2'-H and C_{18'}-H and also for C_2'-H and C_{18'}-H, proposed to belong to two separate helical conformations (16). We confirmed the helical ZZZsss conformation of phycocyanobilin in pyridine but from the NOE enhancement observed for C_{18'}-H and C_{13}-H (not shown). Interestingly, full protonation of 2,3-dihydrobilindiones was reported not to change.

**FIGURE 4** $^{13}$C-NMR spectroscopy of Pr and Pfr states. The Pr state was obtained in pure form after saturating illumination with >705 nm far red light. A mixture of Pfr and Pr states was obtained after illumination of the concentrated sample in a thin capillary with red light, 640 nm.
the $\text{ZZZssa}$ helical conformation as determined from rotating frame Overhauser effect spectroscopy (ROESY) experiments (14), whereas full protonation was suggested to induce extended structures such as observed in protein-bound forms (52). Apparently, remaining chromophore-protein interactions in the chromopeptide stabilize the $\text{ZZZssa}$ conformation, but this is not necessarily taken as evidence for the conformation in the intact receptor. NMR experiments with the chromopeptide in the first instance substantiate assignments (Fig. 1) and characterize conformational exchange reactions (Fig. 2).

Additionally, the $\text{ZZZssa}$ conformation gives confidence that chemical shift differences are not likely to arise from gross configurational differences between chromopeptide and intact protein, assuming similar structures in Cph1 and $D_r$BphP (1). Considering the observation that Cph1, like most phytochromes, relaxes to Pr in the dark in addition to the blue-shifted absorption of the chromopeptide and the $\text{ZZZssa}$ chromophore structure in the $D.\text{radiodurans}$ bacteriophytochrome $D_r$BphP in the Pr state, it is assumed that the Cph1 chromophore is in a Pr-like state (Table 1).

**Chromophore conformation and light-induced changes in the intact Cph1-N515 protein**

The $^{13}\text{C}_4$-resonance shows the largest reversible change in frequency with phototransformation in the intact protein (Fig. 4), which suggests that bond angle changes occur close to $C_4$. In the more upfield region near 95 ppm, where the $C_4$ and $C_5$ resonances are observed, less pronounced changes are visible (Fig. 4). These are interpreted to show an intensity change of the $C_5$-resonance, leaving the $C_5$-resonance mostly unchanged. This view would also fit with the observed changes at $C_4$, which shares $\pi$-orbital valence electrons with $C_5$. Recent evidence suggests that the initial photoisomerization occurs at the $C_{15}$-$C_{16}$ bond (24–26,53). One study using sterically locked biliverdin derivatives implied a Z-anti and E-syn conformation for the $C_{15}$-carbon of the Pr and Pfr states, respectively (53), which has been confirmed for the Pr state of the biliverdin chromophore of $D.\text{radiodurans}$ $D_r$BphP (1). Persuasive evidence for $C_{15}$-$C_{16}$ bond photoisomerization is the lack of phototransformation and high fluorescence quantum yield of a Y167H mutant of Cph1 (27), considering that the conserved tyrosine 167 at that position in the homologous $D.\text{radiodurans}$ Bph is in 4 Å distance of the D-ring (1). Raman spectroscopy studies and mode calculations of photochromobilin containing oat phytochrome (24,25) and biliverdin-containing Agp I bacteriophytochrome (26), both concluded that the Pr to Pfr transformation is initiated by a $\text{ZZZssa}$ to ZZEasa photoisomerization followed by a partial anti to sym thermal $C_5$-$C_6$ bond rotation. We note that the NMR data independently suggest bond angle changes at $C_5$.

Ab initio isotropic chemical shielding calculations were performed for $\text{ZZZssa}$, ZZEasa, and ZZEasa chromophore models in vacuum (Table 2). The GIAO calculations consistently indicated that in energy-minimized conformations a $C_5$-anti to -syn rotation is expected to lead to increased shielding of the $C_4$-carbon atom, in both the $C_4$-$E$ and $C_4$-$Z$ configurations (Table 2). This was also confirmed at the GIAO DFT B3LYP 6-311G+(2d,2p), GIAO HF 6-311G+(2d,2p), and CSGT B3LYP cc-PVDZ levels as well as with solvent reaction field modeling using the polarizable continuum method (37). The calculations performed at different levels of theory all indicated similar changes of the $^{13}\text{C}_4$-resonance frequency resulting from $C_4$-$C_5$-$C_6$-$N_2$ dihedral angle changes. We note that the absolute values of calculated shielding values do not identify conformations, but the differences calculated with bond angle changes are interpreted.

$C_{15}$ Z-E photoisomerization is calculated to lead an ~2-ppm downfield shift of the $C_{15}$-resonance (Table 2), which was also confirmed for geometry-optimized $\text{ZZZssa}$

**TABLE 2** Conformational energies, optical transition energies, and isotropic $^{13}\text{C}$-NMR shielding values calculated for several chromophore models

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Dihedral angle restraints used in the geometry optimization are listed, and none were used if not listed. Conformational energies are given relative to the lowest energy $\text{ZZZssa}$ model. Optical transition energies as calculated by TDDFT are given in eV, with corresponding wavelength, including the oscillator strengths. NMR shielding is reported for all $^{13}\text{C}$-labeled chromophore atoms, relative to TMS (ppm). 1) Conformation based on the $D.\text{radiodurans}$ Bph x-ray structure; 2) ZZEasa structure based on 1; 3) structure shown in Fig. 6 C; and 4) structure shown in Fig. 6 D.
and ZZEssa conformations in vacuum as well as for ZZEssa and ZZEssa phycocyanobilin conformations based on the D. radiodurans Bph x-ray structure, by including a 204° N₃-C₁₄-C₁₅-C₁₆ dihedral angle restraint (Table 2). N₃-C₁₄-C₁₅-C₁₆ dihedral angle changes would lead to further, more pronounced, chemical shift changes of C₁₅ (not shown). TDDFT calculations show that C₁₅ Z-E isomerization could be responsible for red-shifted absorption of the photoprodut but could explain neither the observed C₄ chemical shift changes (not considering possible environmental rearrangements near ring A) nor the absence of chemical shift changes of C₁₅ (Table 2). The NMR data and calculations can therefore not easily be reconciled with a C₁₅ Z-E isomerization in the Pr to Pfr photoreaction without additional low order bond rotation(s) at C₅ and possibly C₁₄. One note of caution concerns the low intensity of the ¹³C₁₅-resonance in the intact protein relative to the chromopeptide, which shows that not the entire population is observed in ¹³C direct experiments comparing Pr and Pfr states in the intact protein (Fig. 2). Our data therefore do not rule out changes at C₁₅, in case its resonance is specifically broadened in the Pfr state as a result of conformational exchange dynamics.

Both fast and slow optical changes in the Pr to Pfr pathway would ideally be reconciled with proposals for the reaction pathway. Notably, the primary photoprodut lumi-R of Pr observed 100 ps after excitation of Cph1 is only slightly red-shifted (8), whereas TDDFT calculations suggest that C₁₅ Z-E isomerization would lead to a considerable red-shift (Table 2). The optical changes occurring during thermal reactions on microsecond and millisecond timescales after excitation of Pr are responsible for the major absorption difference between Pr and Pfr of Cph1 (7,22), implying that these occur as a result of low order bond rotation(s).

A scan of the C₄-C₅-C₆-N₂ dihedral angle in both the ZZEssa and EEZZsa was performed, with constrained geometry optimization for each configuration, to compute the ¹³C-NMR and optical properties (Fig. 5). These calculations identify a region in the ZZEssa (as well as in the EEZZsa) geometry between 275° and 360° (Fig. 6 C) where a decrease of the TDDFT excitation energy is correlated with the deshielding of C₄ (Fig. 5, B and C). In one possible model, C₁₅ Z-E photoisomerization followed by C₄-C₅-C₆=N₂ dihedral angle rotation between 275° and 360°, or by relaxation of the stretched conformation by reduction of the C₄-C₅-C₆ bond angle, might explain the NMR results and possibly the optical and kinetic properties. This reaction model would be very similar to the reaction model proposed on the basis of Raman spectroscopy (24–26). However, the apparent absence of C₁₅ chemical shift differences and the calculated red-shift of the primary photoprodut are not strongly supportive of this possibility, although conformational exchange and environmental effects may play a role in the NMR and optical properties, respectively.

Alternatively, photoisomerization could occur at C₄, followed by C₄-C₅-C₆-N₂ dihedral angle rotation. C₄ Z-E photoisomerization with a 275° dihedral angle leads to only very small optical changes, which would be consistent with the slightly red-shifted primary photoprodut lumi-R of Pr observed 100 ps after excitation of Cph1 (8). A thermal activation barrier between 275° (syn) and 150° (anti) conformations subsequently might separate the lumi-R and the Pfr states, which could be consistent with the red-shifted reaction products which are formed on the microsecond and millisecond timescales after excitation of Pr (7,22). The Z-E isomerization and thermal bond rotation together would constitute a hula-twist motion, which would be more likely given the constraint of covalent attachment of ring A. Cryotrapping of the first metastable “meta-Ra” intermediate of Pr occurs at 233 K (23), which would be consistent with the existence of a rotational barrier in the reaction pathway. Fluorescence measurements of Cph1 at low temperature indicated that the primary photochemical reactions were inhibited below 170 K (51), which together with the low photochemical quantum yield of phototransformation at ambient temperature indicates the presence of a substantial barrier for the initial photoisomerization reaction. Such a barrier may be the result of conformational restraint of the chromophore via covalent linkage on ring A close to the isomerization site.

The models including specific C₄-C₅-C₆=N₂ dihedral angle changes do not use the conformations with the lowest possible conformational energies as optimized and computed in vacuum in the absence of specific interactions (Fig. 5). The associated energy as determined by DFT calculations is reasonable. In addition, the ZZEssa biliverdin in the x-ray structure of D. radiodurans Bph is present in a higher energy conformation, considering the 204° N₃-C₁₄-C₁₅-C₁₆ dihedral angle and the 130° and 135° methine C₅ and C₁₀ bridge angles. DFT geometry optimization indicated that almost kJ/mol is associated with the stretched conformation as refined from the x-ray data increasing the C₅ and C₁₀ methine bond angles by more than 10° and 8 kJ/mol with the twisted N₁-C₁₄-C₁₅-C₁₆ dihedral angle. These calculations assume full protonation on all nitrogens also in the case of D. radiodurans Bph biliverdin. This stretching also causes chemical shift and optical differences. Geometry optimization using redundant coordinates for the 130° and 135° methine C₅ and C₁₀ bridge angles indicates a blue-shifted absorption from subsequent TDDFT calculations. Similarly, intermediate configurations taken from the optimization indicate that stretching could be associated with 3 ppm increased shielding of C₄ and a 0.11 eV increase of the TDDFT excitation energy. This stretching, possibly only locally at C₅, could therefore produce similar NMR and optical changes as C₄-C₅-C₆=N₂ dihedral angle rotation between 275° and 360°. It is possible that conformational stretching and relaxing, rather then low order bond rotations, contribute to the observed NMR and optical changes but in the absence of further molecular information on the Pfr state is not explicitly considered.
FIGURE 5 Molecular properties with C$_4$-C$_5$-C$_6$ dihedral angle changes. A relaxed C$_4$-C$_5$-C$_6$-N$_2$ dihedral angle scan was performed of the ZZZ(s)sa (A–C) (●) and EZZ(s)sa (D–F) (▲) geometries. DFT conformational energies (kJ/mol) (A and D) are given relative to the lowest conformation. Isotropic chemical shielding values are given for the $^{13}$C$_4$-carbon relative to TMS (ppm) (B and E). TDDFT optical transition energies computed for the pure HOMO-LUMO transition (eV) (C and F).
CONCLUSIONS

The reaction models that are discussed aim to satisfy the NMR measurements, as well as the optical and kinetic properties of the phototransformation of Cph1. These neglect effects of specific protein interactions with the chromophore but could provide a generally valid mechanism for the light-induced changes in phytochromes, which have common spectroscopic characteristics despite different polypeptide sequences. The chromophore heterogeneity in the intact receptor domain, which is apparent from the exchange broadening and which is also suggested from transient absorption and fluorescence studies (8,51), adds additional complexity. Evidence favoring C$_{15}$ Z-E photoisomerization is taken from the recent x-ray structure of the D. radiodurans Bph fragment together with the fluorescent Y167H mutant of Cph1. To satisfy NMR and optical properties, additional C$_4$-C$_5$-C$_6$-$N_2$ dihedral angle or possibly C$_4$-C$_5$-C$_6$ bond angle changes are expected, supporting details from previous models based on Raman spectroscopy (24–26). C$_{15}$ Z-E photoisomerization would be expected to lead to rapid optical changes and NMR frequency changes of C$_{15}$, both of which are not observed and would have to be explained by environmental tuning effects or twisted $\pi$-bond geometry and Pfr state-specific conformational exchange, respectively. Alternatively, a C$_4$ Z-E photoisomerization and a C$_5$ syn-anti bond rotation could explain the data and might be at the basis of the photoreaction of the cyanobacterial phytochrome Cph1 and also other (bacterio)phytochromes. Since the chromophore is covalently bound to the protein via the C$_3$, carbon on ring A, hula-twist motions of the C$_4$=C$_5$ and C$_5$-C$_6$ bonds are perhaps more likely, which future calculations may address.

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NMR and DFT Calculations of Cph1

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